

羧酸桥联的四核镧系簇合物的磁热效应和慢磁弛豫

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摘要: 在水热条件下,通过使用羧酸和螯合配体得到了一个系列的四核镧系簇合物,即 $[\text{Ln}_4(\text{mnba})_{12}(\text{tzip})_2(\text{H}_2\text{O})_2]$ ($\text{Ln}=\text{Gd}$ (**1**), Tb (**2**), Er (**3**); Hmnba =间硝基苯甲酸; $\text{tzip}=2-(1H-1,2,4\text{-三唑-3-基})\text{吡啶}$)。这 3 个化合物是同构的,且具有线性的四核簇结构。磁性研究表明,化合物 **1** 和 **3** 中簇内镧系离子之间是弱铁磁耦合的,但化合物 **2** 中铽离子之间是弱的反铁磁相互作用和(或)铽离子激发的斯塔克能级的去布居。化合物 **1** 具有较大的磁热效应($-\Delta S_{\text{m}}^{\text{max}}=20.6 \text{ J}\cdot\text{kg}^{-1}\cdot\text{K}^{-1}$)。交流磁化率测试表明化合物 **3** 展现出频率和温度依赖的虚部信号,这是慢磁弛豫的典型特征,原因是铽离子的强各向异性和铁磁耦合的存在。

关键词: 羧酸; 镧系簇合物; 磁热效应; 慢磁弛豫

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Carboxylate-Bridged Tetranuclear Lanthanide Clusters: Magnetocaloric Effect and Slow Magnetic Relaxation

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Abstract: By using carboxylate and chelating ligands, a family of tetranuclear lanthanide clusters, namely $[\text{Ln}_4(\text{mnba})_{12}(\text{tzip})_2(\text{H}_2\text{O})_2]$ ($\text{Ln}=\text{Gd}$ (**1**), Tb (**2**) and Er (**3**), Hmnba =m-nitrobenzoic acid, $\text{tzip}=2-(1H-1,2,4\text{-triazol-3-yl})\text{pyridine}$), has been obtained under hydrothermal conditions. The three complexes exhibit linear tetranuclear clusters bridged by carboxylates with *syn, syn- μ_2 - η^1 : η^1* mode. Magnetic investigation indicates weak ferromagnetic interaction between adjacent Gd^{III} or Er^{III} ions of the Ln_4 cluster in **1** and **3**, while weak intra-molecular antiferromagnetic interaction between Tb^{III} ions and/or depopulation of the Tb^{III} excited Stark sub-levels in **2**. Complex **1** exhibits a significant magnetocaloric effect with $-\Delta S_{\text{m}}^{\text{max}}=20.6 \text{ J}\cdot\text{kg}^{-1}\cdot\text{K}^{-1}$ and ac susceptibility measurements reveal frequency- and temperature-dependent out-of-phase signal under 5 kOe dc field in **3**, being typical slow magnetic relaxation behavior due to strong anisotropy of Er^{III} and ferromagnetic coupling. CCDC: 978830, **1**; 978831, **2**; 978832, **3**.

Key words: carboxylate; lanthanide clusters; magnetocaloric effect; slow magnetic relaxation

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0 Introduction

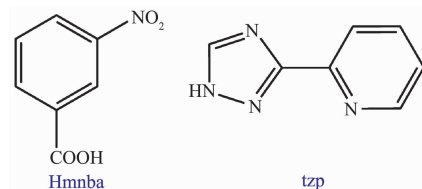
The investigation of lanthanide (Ln) clusters has recently become an active field for their both fascinating structures and exceptional applications as molecular coolers and single-molecule magnets (SMMs)^[1-6]. On one hand, Gd^{III} clusters could be regarded as candidate materials for magnetic refrigerators because of negligible magnetic anisotropy (D), large spin ground state (S) and low-lying excited spin states of Gd^{III} ion and weak couplings between Gd^{III} ions^[7-11]. Generally, the entropy change ($-\Delta S_m$) is employed to represent the magnetocaloric effect (MCE) of molecular magnetorefrigerants^[12-13]. The magnetic intensity (M_w/N_{Gd}) and magnetic interaction (θ) between Gd^{III} ions are proposed to be main factors to affect MCE for the Gd-type magnetic refrigerants^[14-16]. To reduce M_w/N_{Gd} ratio and $|\theta|$ value of Gd^{III} complexes, the utilization of light ligands to synthesize Gd^{III} clusters supply an effective tool.

On the other hand, Ln-based SMMs with large energy barriers have been a hot research topic for molecular magnets compared with $3d/3d-4f$ based SMMs^[17-19]. Ln^{III} clusters (especially for Tb^{III}, Dy^{III}, Ho^{III} and Er^{III} types) have recently become favorable candidates to explore SMMs, since the S and D of Ln^{III} ions could lead to an relatively large anisotropic energy barrier (U_{eff}) that prevents the reversal of the molecular magnetization^[20-21].

Till now, most of Ln-clusters were constructed from Schiff-base and calix[4]arenes ligands with their chelating characteristic^[22-25]. The mixed-ligand strategy, especially the utilization of carboxylates and N-donor ligands, has been employed to construct discrete clusters and coordination polymers as a powerful synthetic approach, while the design and synthesis of discrete Ln-clusters with unique structures and magnetic properties still remain a great challenge because of different affinities and coordination capabilities of the Ln^{III} ions to O-donors and N-donors^[26-27].

As an extension of our studies on the synthesis and magnetic investigation of Ln^{III} complexes^[28-31], herein, we choose sterically hindered Hmnba (m-nitrobenzoic acid) and corner ligands 2-(1*H*-1,2,4-triazol-3-yl)

pyridine) (tzp) to construct low-dimensional structures (Scheme 1). Fortunately, a series of tetranuclear Ln^{III} clusters, namely $[Ln_4(mnba)_{12}(tzip)_2(H_2O)_2]$ (Ln=Gd (**1**), Tb (**2**) and Er (**3**)) were successfully synthesized.



Scheme 1 Ligands used for the synthesis of **1**~**3**

Magnetic analyses reveal that complex **1** is weakly ferromagnetic coupled with $-\Delta S_m^{max}=20.6 \text{ J} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}$ for $\Delta H=7 \text{ T}$ at 2.0 K and complex **3** displays slow relaxation of the magnetization. Strong quantum tunnelling effect excludes the existence of slow magnetic relaxation for **2** although 2 kOe dc field was exerted.

1 Experimental

1.1 Materials and instrumentation

All chemicals were of reagent grade and used as purchased without further purification. Elemental analysis (C, H and N) was performed on a Perkin-Elmer 240C analyzer (Perkin-Elmer, USA). The X-ray powder diffraction (PXRD) spectra were recorded on a Rigaku D/Max-2500 diffractometer at 60 kV, 300 mA for a Cu-target tube and a graphite monochromator. Simulation of the PXRD spectra were carried out by the single-crystal data and diffraction-crystal module of the Mercury (Hg) program available free of charge via the Internet at <http://www.iucr.org>. IR spectra were measured in the range of 400~4 000 cm^{-1} on a Tensor 27 OPUS FT-IR spectrometer using KBr pellets (Bruker, German). Magnetic data were measured by a Quantum Design MPMS-XL-7 SQUID magnetometer. Diamagnetic corrections were estimated by using Pascal constants and background corrections by experimental measurement on sample holders.

1.2 Preparation of **1**~**3**

$[Gd_4(mnba)_{12}(tzip)_2(H_2O)_2]$ (**1**): A mixture of Gd_2O_3 (181 mg, 0.5 mmol), Hmnba (334 mg, 2 mmol) and tzp (66.6 mg, 0.5 mmol) in 10 mL H_2O was sealed in a Teflon-lined autoclave and heated to 160 $^{\circ}\text{C}$ for 2

days. After the autoclave was cooled to room temperature in 12 h, Cubic colorless crystals were collected with 30% yield based on Gd^{III}. Anal. Calcd. for C₉₈H₆₄O₅₀N₂₀Gd₄ (%): C, 39.89; H, 2.19; N, 9.49. Found(%): C, 39.78; H, 2.78; N, 9.35. IR (KBr, cm⁻¹): 3 579m, 3 504m, 3 161m, 3 087s, 2 901m, 2 765w, 1 580 s, 1 483s, 1 344s, 1 263s, 1 166m, 1 080s, 995w, 912m, 831m, 788s, 725s, 651m, 576w, 523w, 416m.

[Tb₄(mnba)₁₂(tbp)₂(H₂O)₂] (**2**): The same procedure as that for **1** was used for this complex except that Gd₂O₃ (181 mg, 0.5 mmol) was replaced by Tb₂O₃ (183 mg, 0.5 mmol) and the holding time is 3 days. Block colorless crystals were collected with ~40% yield based on Tb^{III}. Anal. Calcd. for C₉₈H₆₄O₅₀N₂₀Tb₄(%): C, 39.80; H, 2.18; N, 9.47. Found(%): C, 40.13; H, 2.89; N, 9.60. IR (KBr, cm⁻¹): 3 494w, 3 157w, 3 080m, 2 893 m, 2 767w, 1 598s, 1 517s, 1 415s, 1 350s, 1 266m, 1 082 m, 995w, 910w, 790m, 723s, 649w.

[Er₄(mnba)₁₂(tbp)₂(H₂O)₂] (**3**): The same procedure as that for **1** was used for this complex except that Gd₂O₃ (181 mg, 0.5 mmol) was replaced by Er₂O₃ (191 mg, 0.5 mmol) and the holding time is 3 days. Block pink crystals were collected with ~30% yield based on Er^{III}. Anal. Calcd. for C₉₈H₆₄O₅₀N₂₀Er₄(%): C, 39.36; H, 2.16; N, 9.37. Found (%): C, 39.64; H, 2.78; N,

9.43. IR (KBr, cm⁻¹): 3 496w, 3 163w, 3 083m, 2 891w, 1 608s, 1 525s, 1 478s, 1 410s, 1 346s, 1 267m, 1 166w, 1 079m, 1 001w, 909w, 830w, 719s, 650m, 584w, 518w, 413w.

1.3 Crystallographic data and structure refinements

The single-crystal X-ray diffraction data of **1~3** were collected on a Rigaku SCX-mini diffractometer at 293(2) K with Mo K α radiation (λ =0.071 073 nm) by ω scan mode. The program CrystalClear^[32] was used for the integration of the diffraction profiles. The structures were solved by direct method using the SHELXS program of the SHELXTL package and refined by full-matrix least-squares methods with SHELXL^[33]. The non-hydrogen atoms were located in successive difference Fourier syntheses and refined with anisotropic thermal parameters on F^2 . All hydrogen atoms of ligands were generated theoretically at the specific atoms and refined isotropically with fixed thermal factors. The hydrogen atoms of water in **1~3** were added by the difference Fourier maps and refined with suitable constrains. A summary of the crystallographic data, data collection, and refinement parameters for **1~3** is provided in Table 1.

CCDC: 978830, **1**; 978831, **2**; 978832, **3**.

Table 1 Crystal data and structure refinements for **1~3**

	1	2	3
Formula	C ₉₈ H ₆₄ O ₅₀ N ₂₀ Gd ₄	C ₉₈ H ₆₄ O ₅₀ N ₂₀ Tb ₄	C ₉₈ H ₆₄ O ₅₀ N ₂₀ Er ₄
Formula weight	2 950.69	2 957.37	2 990.73
Crystal system	Triclinic	Triclinic	Triclinic
Space group	$P\bar{1}$	$P\bar{1}$	$P\bar{1}$
a / nm	1.068 7(2)	1.063 5(2)	1.060 2(2)
b / nm	1.401 2(3)	1.392 3(3)	1.392 5(3)
c / nm	1.993 3(4)	1.977 3(4)	1.972 4(4)
α / (°)	70.33(3)	70.49(3)	71.06(3)
β / (°)	86.67(3)	86.71(3)	87.15(3)
γ / (°)	76.42(3)	76.40(3)	76.48(3)
V / nm ³	2.731 4(9)	2.681 5(9)	2.676 9(9)
D_c / (g·cm ⁻³)	1.794	1.831	1.855
Z	1	1	1
$F(000)$	1 448	1 452	1 464
μ / mm ⁻¹	2.504	2.714	3.212
Collected reflections	28 415	23 302	28 489

Continued Table 1

Unique reflections	12 390	9 453	12 231
R_{int}	0.036 5	0.042 3	0.046 7
$R_1^a / wR_2^b [I > 2\sigma(I)]$	0.038 7 / 0.073 1	0.043 4 / 0.094 4	0.044 6 / 0.068 6
GOF on F^2	1.081	1.132	1.063

$$^a R = \sum (|F_o| - |F_c|) / \sum |F_o|; ^b wR = [\sum w(|F_o|^2 - |F_c|^2)^2 / (\sum w|F_o|^2)]^{1/2}$$

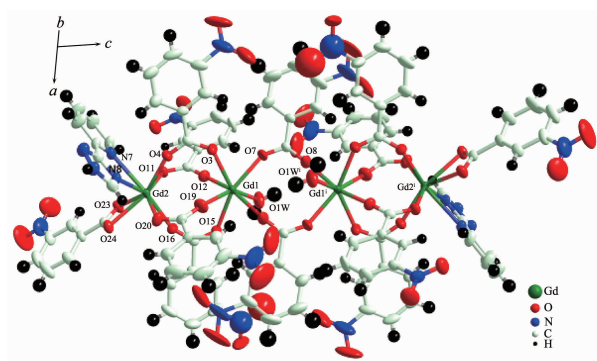
2 Results and discussion

2.1 Synthesis

The mix-ligand strategy has been employed to construct the linear Ln-clusters successfully. As effective terminal co-ligands, tzp plays a key role in the formation of discrete lanthanide clusters. Compared with lanthanide salts, the use of Ln_2O_3 provides not only a slow-release Ln^{III} ion source but also a pH regulator of the reactions.

2.2 Description of crystal structures

Single crystal X-ray analysis reveals that **1**~**3** are isostructural, thus only the structure of **1** is described in detail. Complex **1** crystallizes in the triclinic space group $P\bar{1}$ with the asymmetric unit including two Gd^{III} ions, six mnba ligands, one tzp ligand and one coordinated H_2O . Gd1 is located in a seven coordinated environment constructed with seven O atoms from one water molecule and six carboxylates. Gd2 center exhibits an eight-coordinated environment with two N atoms of one tzp ligand and six carboxylate O atoms from five mnba ligands. Gd1 and Gd2 are connected by four carboxylate groups with *syn, syn- μ_2 - η^1 : η^1* to form a linear Gd_4 cluster (Fig.1).



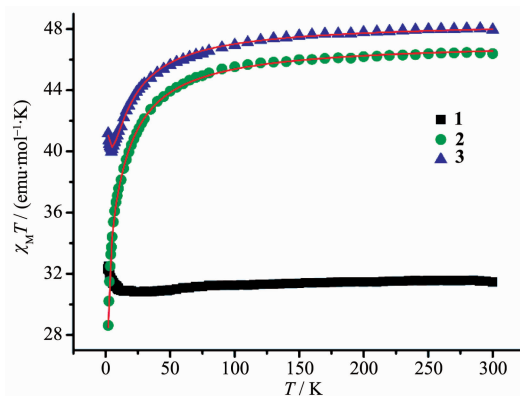
Symmetry code: $x+1, -y+1, -z+2$

Fig.1 View of the molecular structure showing 30% probability thermal ellipsoids of **1**

2.3 Magnetic studies

The magnetic properties of **1**~**3** were studied by solid state magnetic susceptibility measurements in 2.0~300 K range at 1 kOe dc field and the isothermal field-dependent magnetizations $M(H)$ at fields up to 70 kOe at 2.0 K. Before the magnetic measurements of **1**~**3**, their crushed crystalline samples were used to measure X-ray powder diffraction (PXRD, Fig.S1, SI) to confirm their phase purities.

Complex **1** contains isotropic $\text{Gd}^{\text{III}}(f^7)$ with a ground state $^8S_{7/2}$, and the first excited state $^6P_{7/2}$ is very high in energy, while complexes **2** and **3** include other anisotropic Ln^{III} ions. Generally, the magnetism of lanthanide (Ln) clusters is very difficult to explain because of the exchange-coupling and large orbital contributions as well as the crystal field perturbation^[34-35]. The magnetic properties in the form of $\chi_M T$ vs T plots of **1**~**3** are shown in Fig.2. The room-temperature $\chi_M T$ products estimated as 31.45 (**1**), 46.39 (**2**) and 47.96 (**3**) $\text{emu} \cdot \text{mol}^{-1} \cdot \text{K}$ are in relative good agreement with the presence of four lanthanide metal ions: four Gd^{III} ions ($S = 7/2, L = 0, J = 7/2, g = 2, C = 7.88 \text{ emu} \cdot \text{mol}^{-1} \cdot \text{K}$) for **1**, four Tb^{III} ions ($S = 3, L = 3, J = 6, g = 3/2, C = 11.82 \text{ emu} \cdot \text{mol}^{-1} \cdot \text{K}$) for **2** and four Er^{III} ions ($S = 3/2, L = 6, J =$



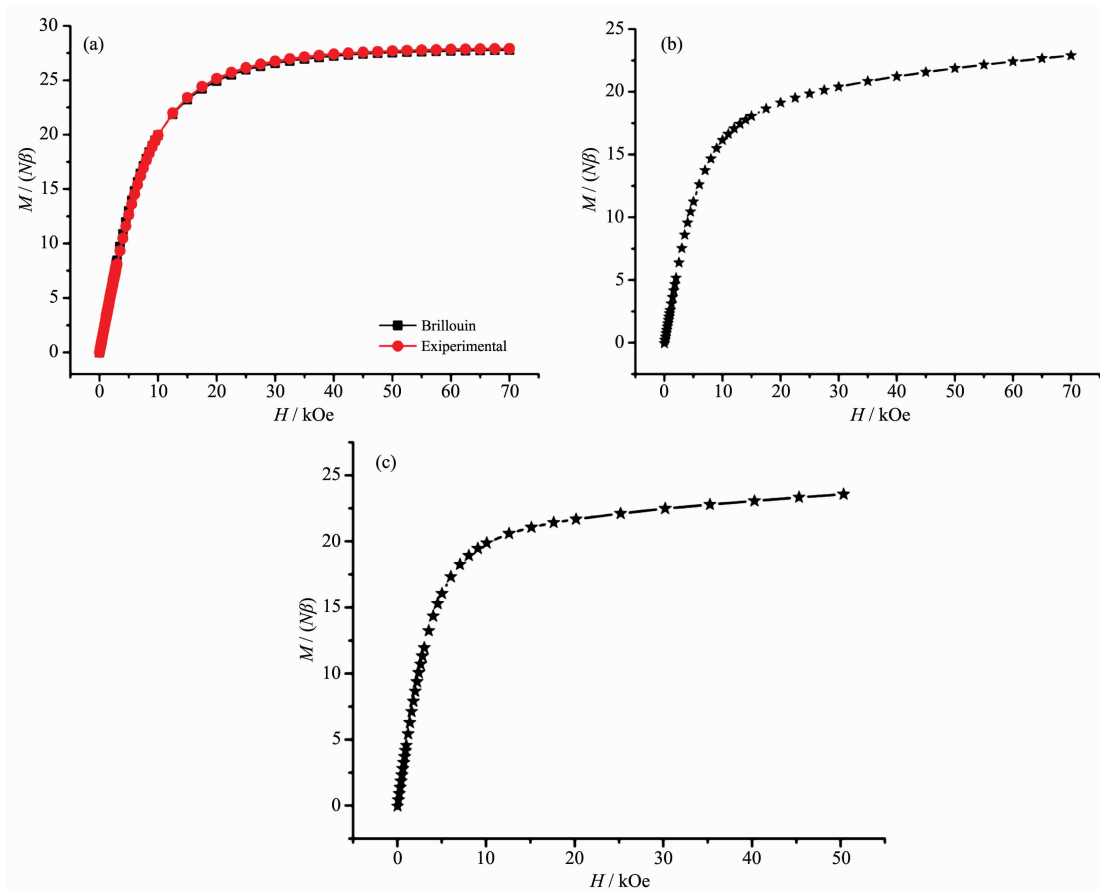
Red lines represent the best fitting

Fig.2 Plots of $\chi_M T$ vs T for **1**~**3**

15/2, $g=6/5$, $C=11.48 \text{ emu} \cdot \text{mol}^{-1} \cdot \text{K}$) for **3**. For **1**, as the temperature decreases, the $\chi_M T$ value stays nearly constant in the high temperature range with a value of $30.86 \text{ emu} \cdot \text{mol}^{-1} \cdot \text{K}$ at 14 K. Upon further cooling the temperature to 2.0 K, $\chi_M T$ abruptly increases to a maximum value ($32.31 \text{ emu} \cdot \text{mol}^{-1} \cdot \text{K}$, indicating the ferromagnetic (F) interaction between Gd^{III} ions in the Gd_4 cluster. For **2**, up lowering of the temperature to 2.0 K, $\chi_M T$ value stays nearly constant at high temperatures, and then decreases sharply to a minimum value ($28.62 \text{ emu} \cdot \text{mol}^{-1} \cdot \text{K}$), which indicates weak antiferromagnetic (AF) interaction in the Tb_4 cluster and/or depopulation of the Tb^{III} excited Stark sub-levels. The Stark sub-levels of the anisotropic Tb^{III} ions may be progressively thermally depopulated leading to a decrease of the $\chi_M T$ value. For **3**, as the temperature decreases, the value of $\chi_M T$ slowly decreases down to a minimum value of $39.96 \text{ emu} \cdot \text{mol}^{-1} \cdot \text{K}$ at 4.5 K. On cooling the temperature to 2 K, $\chi_M T$ abruptly

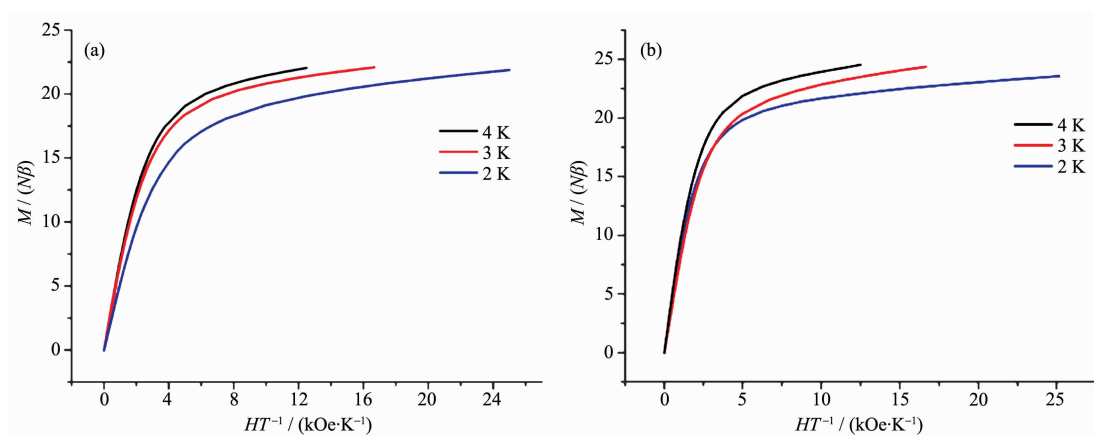
increases to the maximum value ($41.17 \text{ emu} \cdot \text{mol}^{-1} \cdot \text{K}$), indicating ferromagnetic coupling between Er^{III} ions in the Er_4 cluster.

The magnetizations slowly increase and tend to a value of $27.77N\beta$ at 70 kOe, and the experimental magnetization plot of **1** is nearly consistent with the red line that presents the Brillouin function for four magnetically uncoupled Gd^{III} ions with $S=7/2$ and $g=2.0$, which further confirms the weak F behavior for **1** as similar literatures^[3-4] (Fig.3a). The field dependences of the magnetizations at 2.0 K for **2** and **3** show rapid increases of the magnetizations at low fields, reaching about $16.13N\beta$ and $19.87N\beta$ at 10 kOe, and linear increases at high fields without achieving a complete saturation at 70 or 50 kOe ($22.90N\beta$ for **2** and $23.57N\beta$ for **3**, Fig.3b and 3c), which could be explained by the fact that the depopulation of the Stark levels of the $\text{Ln}^{\text{III}} 2S+1L_J$ ground state under the ligand-field perturbation produces a much smaller effective spin.



Red solid line represents Brillouin function of four magnetically isolated Gd^{III} ions with $S=7/2$ and $g=2.0$

Fig.3 M vs H curves of **1** (a), **2** (b) and **3** (c) at 2.0 K

Fig.4 Curves of M vs H/T for **2** (a) and **3** (b)

For **2** and **3**, the M vs H/T (Fig.4) data at 2~4 K shows non-superposition plots and a rapid increase of the magnetization at low fields without any sign of saturation at 50 kOe. The reason is most likely because of anisotropy and important crystal-field effect at the Tb^{III} or Er^{III} ions, which eliminates the degeneracy of the 7F_6 and $^4I_{15/2}$ ground states. Reduced magnetization curves do not superimpose, further indicating the presence of a significant magnetic anisotropy and/or low lying excited states^[15].

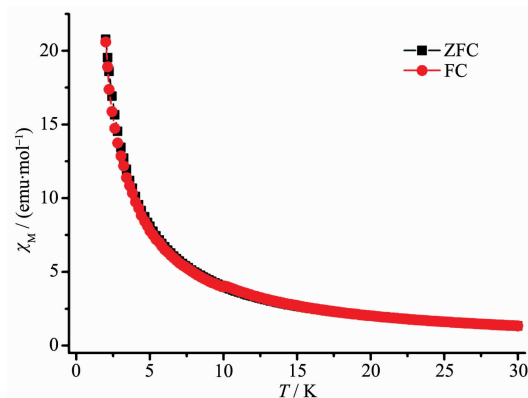
The recently developed non-critical scaling theory could be used to study the F/AF behaviors of **2** and **3** based on the sum of two exponential functions (as shown in Eq.1)^[36-38].

$$\chi T = A \exp(E_1/T) + B \exp(E_2/T) \quad (1)$$

In Eq.1, $A+B$ is the high-temperature extrapolated Curie constant and E_1 and E_2 denote the magnitude of the intracluster magnetic interaction. The first term in Eq.1 represents an F/AF contribution to the moment that is dominant at low temperatures, whereas the second term reflects the crystal-field effect because the interionic interactions between the internal $4f$ electrons are usually very weak. The best fit of the experimental data gives that $A+B=47.22$ emu \cdot mol⁻¹ \cdot K, $E_1=-0.61$ K and $E_2=-20.94$ K for **2** and $A+B=48.54$ emu \cdot mol⁻¹ \cdot K, $E_1=0.08$ K and $E_2=-18.12$ K for **3** (Fig.2). The small values of E_1 of **2** and **3** further indicate very weak magnetic interactions between the Tb^{III}/Er^{III} ions, which is in good agreement with the prediction that the Ln-Ln interaction is expected to be very weak, due to the shielding of the f-orbitals and

the consequent poor overlap with the bridging ligand orbitals^[12].

To characterize the low-temperature behaviors of **1**, the temperature dependencies of field-cooled (FC) and zero-field-cooled (ZFC) magnetization were measured under a field of 50 Oe upon warming from 2.0 K (Fig.5). The FC curve coincides with the ZFC curve and the magnetizations increase monotonically with the decrease of temperature, and no maximum is observed. These results indicate that **1** does not exhibit magnetic ordering above 2.0 K.

Fig.5 FC/ZFC curves at 2~30 K for **1**

Considering the weak magnetic couplings between the Gd^{III} ions and potential application of Gd^{III} complexes for magnetic refrigeration, we investigated the magnetocaloric properties of **1**. We used the magnetic entropy change (ΔS_m) to evaluate MCE, which could be calculated by the Maxwell equation ($\Delta S_m(T)_{\Delta H} = \int [\partial M(T, H) / \partial T]_H dH$)^[39-41]. According to the equation, we could obtain the $-\Delta S_m$ from the experimental magnetization data

(Fig.6a), and the curves of $-\Delta S_m$ are depicted in Fig. 6b. The obtained $-\Delta S_m^{\max}$ gives the value of $20.6 \text{ J} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}$ (the theoretical $-\Delta S_m^{\max}$ is $23.4 \text{ J} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}$ calculated with $-\Delta S_m^{\max} = 4R \ln(2S+1)$, R is the gas constant) for a field change of 7 T at 2.0 K. If $-\Delta S_m^{\max}$ is given per unit of volume, it is equivalent to $36.96 \text{ mJ} \cdot \text{cm}^{-3} \cdot \text{K}^{-1}$. Although various discrete Gd^{III} clusters have been constructed, their magnetocaloric properties have rarely been reported. Previous literatures report only twelve Gd^{III} clusters with significant MCE ($-\Delta S_m^{\max} > 20 \text{ J} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}$), as shown in Table 2.

To investigate possible SMM behaviors of **2** and

3, alternating current (ac) susceptibility measurements were carried out in the temperature range of 15~2.0 K under $H_{\text{dc}}=0 \text{ Oe}$ and $H_{\text{ac}}=3.5 \text{ Oe}$ for variable frequencies (from 1 488 to 10 Hz). Unfortunately, although all the in-phase curves (χ') are almost consistent without peaks, there is no frequency dependent out-of-phase signal even up to 997 Hz (Fig.S2a and S2c). In order to weaken the quantum tunneling effect, 2 kOe dc field were applied to further study the dynamic properties. The ac signal of **2** is still poor and slow magnetic relaxation is not observed (Fig.S2b), while there is weak frequency dependent out-of-phase signal for **3**

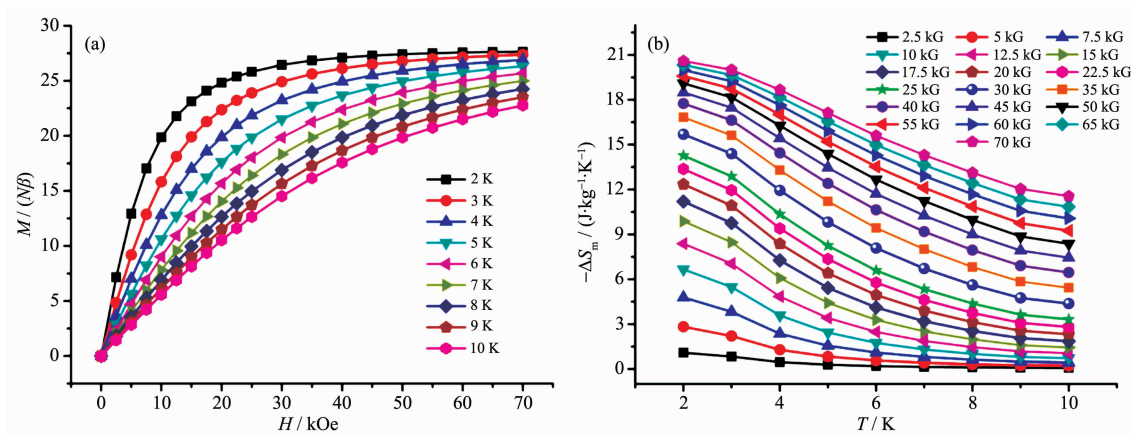


Fig.6 (a) M vs H curves of **1** at $T=2\sim 10 \text{ K}$ and $H=2.5\sim 70 \text{ kOe}$; (b) Experimental $-\Delta S_m$ obtained from magnetization data of **1** at different fields and temperatures

Table 2 Comparison of $-\Delta S_m^{\max}$ (larger than $20.0 \text{ J} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}$) among **1** and Gd^{III} clusters associated with potential molecule-based magnetic coolers*

Complex	M_w / N_{Gd}	Magnetic interaction (θ / K)	$-\Delta S_m^{\max} / (\text{J} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}) (\Delta H)$	$-\Delta S_m^{\max} / (\text{mJ} \cdot \text{cm}^{-3} \cdot \text{K}^{-1})$
$[\text{Gd}_{104}]^{[42]}$	292.44	AF (−4.11)	46.9 (7 T)	138.12
$[\text{Gd}_{34}]^{[10]}$	340.40	AF (−0.16)	46.12 (7 T)	89.98
$[\text{Gd}_{48}]^{[11]}$	313.83	AF (−3.57)	43.6 (7 T)	120.7
$\{[\text{Gd}(\text{OAc})_3(\text{H}_2\text{O})_2]_2\} \cdot 4\text{H}_2\text{O}^{[13]}$	406.44	F (0.32)	41.6 (7 T)	82.78
$[\text{Gd}_{38}]^{[11]}$	376.27	AF (−2.99)	37.9 (7 T)	102.0
$[\text{Gd}_4(\text{OAc})_4(\text{acac})_8(\text{H}_2\text{O})_4]^{[12]}$	432.53	F (0.23)	37.7 (7 T)	70.24
$[\text{Gd}_{10}]^{[43]}$	463.59	AF	37.4 (7 T)	43.01
$[\text{Gd}_6]^{[44]}$	434.65	F	33.5 (7 T)	56.68
$[\text{Gd}_3]^{[44]}$	572	AF	31.3 (7 T)	64.99
$[\text{Gd}_{10}(3\text{-TCA})_{22}(\mu_3\text{-OH})_8(\text{H}_2\text{O})_4]^{[30]}$	457.76	AF (−1.78)	31.22 (7 T)	68.64
$[\text{Gd}_4]^{[45]}$	632.19	AF	27.2 (7 T)	40.96
$[\text{Gd}_2(\text{OAc})_2(\text{Ph}_2\text{acac})_6(\text{MeOH})_2]^{[12]}$	694.81	F (0.18)	23.7 (7 T)	36.43
$[\text{Gd}_7]^{[9]}$	549.09	AF	23 (7 T)	41.33
$[\text{Gd}_4(\text{mnba})_{12}(\text{tzp})_2(\text{H}_2\text{O})_2]$ (1)	737.67	F (0.51)	20.6 (7 T)	36.96

* $-\Delta S_m^{\max} / (\text{mJ} \cdot \text{cm}^{-3} \cdot \text{K}^{-1}) = [-\Delta S_m^{\max} / (\text{J} \cdot \text{kg}^{-1} \cdot \text{K}^{-1})] \cdot [D_c / (\text{g} \cdot \text{cm}^{-3})]$

(Fig.S2d). Then 5 kOe dc field was exerted and attempted to obtain better ac signals. Therefore, the peaks can be observed obviously both in χ_M' and χ_M'' curves (see Fig.7), which suggested the existence of slow magnetic relaxation behavior in **3**. As aforementioned, strong anisotropy of Er^{III} ions and weak ferromagnetic interaction presumably lead to the field-induced slow magnetic relaxation behavior in **3**.

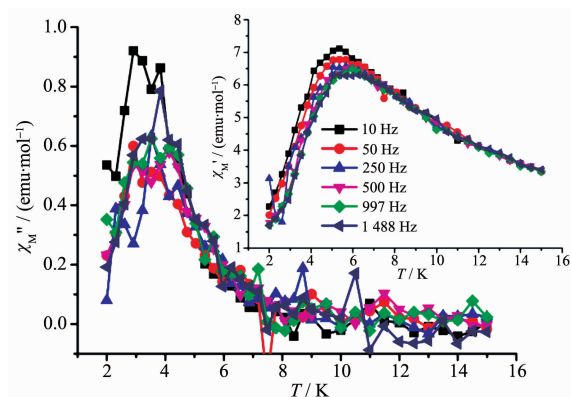


Fig.7 Temperature dependence of the ac χ_M at different frequencies for **3** with $H_{dc}=5$ kOe

3 Conclusions

A type of linear tetranuclear lanthanide clusters (**1**~**3**) constructed from the monocarboxylate and terminal co-ligand has been synthesized in hydrothermal reactions. Magnetic investigation indicates that **1**~**3** are weakly coupled with **1** displaying large MCE with $-\Delta S_m^{\text{max}}=20.6 \text{ J} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}$ and **3** exhibiting slow magnetic relaxation behavior for the strong anisotropy and ferromagnetic contribution. Complex **2** does not show slow magnetic relaxation behavior because of weak antiferromagnetic interaction and strong quantum tunneling effect, although 2 kOe dc field was exerted.

References:

- [1] Zheng Y Z, Zhou G J, Zheng Z, et al. *Chem. Soc. Rev.*, **2014**, **43**:1462-1475
- [2] Sharples J W, Collison D. *Polyhedron*, **2013**, **54**:91-103
- [3] Woodruff D N, Winpenny R E P, Layfield R A. *Chem. Rev.*, **2013**, **113**:5110-5148
- [4] Liu J L, Chen Y C, Guo, F S, et al. *Coord. Chem. Rev.*, **2014**, **281**:26-49
- [5] Rinehart J D, Long J R. *Chem. Sci.*, **2011**, **2**:2078-2085
- [6] Habib F, Murugesu M. *Chem. Soc. Rev.*, **2013**, **42**:3278-3288
- [7] Wang P, Shannigrahi S, Yakovlev N L, et al. *Chem. Asian J.*, **2013**, **8**:2943-2946
- [8] Liu S J, Zhao J P, Tao J, et al. *Inorg. Chem.*, **2013**, **52**:9163-9165
- [9] Sharples J W, Zheng Y Z, Tuna F, et al. *Chem. Commun.*, **2011**, **47**:7650-7652
- [10] Chang L X, Xiong G, Wang L, et al. *Chem. Commun.*, **2013**, **49**:1055-1057
- [11] Guo F S, Chen Y C, Mao L L, et al. *Chem. Eur. J.*, **2013**, **19**:14876-14885
- [12] Guo F S, Leng J D, Liu J L, et al. *Inorg. Chem.*, **2012**, **51**:405-413
- [13] Evangelisti M, Roubeau O, Palacios E, et al. *Angew. Chem. Int. Ed.*, **2011**, **50**:6606-6609
- [14] Liu S J, Xie C C, Jia J M, et al. *Chem. Asian J.*, **2014**, **9**:1116-1122
- [15] Lorusso G, Sharples J W, Palacios E, et al. *Adv. Mater.*, **2013**, **25**:4653-4656
- [16] Han S D, Miao X H, Liu S J, et al. *Inorg. Chem. Front.*, **2014**, **1**:549-552
- [17] Blagg R J, Murn C A, McInnes E J L, et al. *Angew. Chem. Int. Ed.*, **2011**, **50**:6530-6533
- [18] Gao F, Cui L, Liu W, et al. *Inorg. Chem.*, **2013**, **52**:11164-11172
- [19] Guo Y N, Xu G F, Wernsdorfer W, et al. *J. Am. Chem. Soc.*, **2011**, **133**:11948-11951
- [20] Jiang S D, Liu S S, Zhou L N, et al. *Inorg. Chem.*, **2012**, **51**:3079-3087
- [21] Jiang S D, Wang B W, Sun H L, et al. *J. Am. Chem. Soc.*, **2011**, **133**:4730-4733
- [22] Rell N M, Anwar M U, Drover M W, et al. *Inorg. Chem.*, **2013**, **52**:6731-6742
- [23] Anwar M U, Thompson L K, Dawe L N, et al. *Chem. Commun.*, **2012**, **48**:4576-4578
- [24] Liu C M, Zhang D Q, Hao X, et al. *Cryst. Growth Des.*, **2012**, **12**:2948-2954
- [25] Xu X, Zhao L, Xu G F, et al. *Dalton. Trans.*, **2011**, **40**:6440-6444
- [26] Liu S J, Zhao J P, Song W C, et al. *Inorg. Chem.*, **2013**, **52**:2103-2109
- [27] Sessoli R, Powell A K. *Coord. Chem. Rev.*, **2009**, **253**:2328-2341
- [28] Jia J M, Liu S J, Cui Y, et al. *Cryst. Growth Des.*, **2013**, **13**:4631-4634
- [29] Liu S J, Zeng Y F, Xue L, et al. *Inorg. Chem. Front.*, **2014**, **1**:200-206
- [30] Han S D, Miao X H, Liu S J, et al. *Chem. Asian J.*, **2014**, **9**:

- 3116-3120
- [31]Miao X H, Han S D, Liu S J, et al. *Chin. Chem. Lett.*, **2014**, **25**:829-834
- [32]Rigaku. *CrystalClear*, Process-Auto Rigaku Americas Corporation, The Woodlands, Texas, **1998**.
- [33]Sheldrick G M. *SHELXL97, Program for Crystal Structure Refinement*, University of Göttingen, Göttingen, Germany, **1997**.
- [34]Wu M F, Wang M S, Guo S P, et al. *Cryst. Growth Des.*, **2011**,**11**:372-381
- [35]Kahn M L, Sutter J P, Golhen S, et al. *J. Am. Chem. Soc.*, **2000**,**122**:3413-3421
- [36]Zheng Y Z, Lan Y H, Wernsdorfer W, et al. *Chem. Eur. J.*, **2009**,**15**:12566-12570
- [37]Souletie J, Rabu P, Drillon M. *Phys. Rev. B*, **2005**,**72**:214427
- [38]Drillon M, Panissod P, Rabu P, et al. *Phys. Rev. B*, **2002**, **65**:104404
- [39]Zheng Y Z, Evangelisti M, Winpenny R E P. *Angew. Chem. Int. Ed.*, **2011**,**50**:3692-3695
- [40]Zheng Y Z, Pineda E M, Helliwell M, et al. *Chem. Eur. J.*, **2012**,**18**:4161-4165
- [41]FAN Shao-Yong(范少勇), HUANG Ke-Di(黄科棣), TANG Qing-Mei(唐清美), et al. *Chinese J. Inorg. Chem.*(无机化学学报), **2014**,**30**:1167-1173
- [42]Peng J B, Kong X J, Zhang Q C, et al. *J. Am. Chem. Soc.*, **2014**,**136**:17938-17941
- [43]Adhikary A, Jena H S, Biswas S, et al. *Chem. Asian J.*, **2014**,**9**:1083-1090
- [44]Adhikary A, Sheikh J A, Biswas S, et al. *Dalton Trans.*, **2014**,**43**:9334-9343
- [45]Sheikh J A, Adhikary A, Konar S. *New J. Chem.*, **2014**,**38**: 3006-3014